TRAPPING OF CARBANIONIC SPECIES SYNTHESIS OF NEW SALT-FREE YLIDS AND PHOSPHORANES

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Summary - Syntheses of new salt-free ylids and phosphoranes by addition of a trivalent phosphorus compound with dimethylacetylene dicarboxylate in presence of a trapping reagent are described. The results are consistent with trapping of carbanionic species.

In a previous note (1), we reported the synthesis of new vinylphosphoranes obtained from an addition reaction between spirophosphoranes which present a P-H bond and mono and dicarbonyl acetylenic compounds. The most likely

Spirophosphorane $\underline{1}$ reacts in its tricoordinated tautomeric form $\underline{1'}$ leading to a carbanionic quasi-phosphonium species $\underline{2a}$, $\underline{2a}$ stabilizes itself by protonation and formation of the phosphorane $\underline{2}$. The tricoordinated tautomeric form $\underline{1'}$ includes both the P III $\underline{0}$ P-0-structure and a protic nucleophilic group-OH. Therefore, in order to extend the scope of this reaction, we have examined the reaction of $\underline{0}$ P-0Me with MeOH and noted that it provides a novel and direct route to the synthesis of the monocyclic phosphoranes $\underline{3}$ and $\underline{4}$ in a nearly quantitative yield (Table I).

The reaction A gives rise to an ylid if it is performed with another tricoordinated phosphorus compound namely $P(OMe)_3$ or $P(NMe)_3$: reaction B.

B
$$(X_3)P$$
 + Me OH + $\begin{pmatrix} COOMe \\ C \\ C \\ COOMe \end{pmatrix}$ $(X)_3P = C \begin{pmatrix} COOMe \\ H \\ COOMe \end{pmatrix}$

 $5 \quad X = OMe \quad \underline{6} \quad X = NMe_2$

On the basis of these results it is clear that the course of the reacting species leading to an ylid or to a phosphorane is controlled by the structure of the trivalent phosphorus compound. This assumption was proved by the following studies. We investigated several reactions using tricoordinated compounds 7'-10', following the same experimental conditions:

All of them led to an ylid analogous to $\underline{5}$ and $\underline{6}$ (Table II). Structure proofs are based on the NMR (1 H, 31 P) spectra analysis (see Table I and II), the IR spectra, combustion analysis and a chemical reaction : hydrolysis. For instance

$$(Me_2N)_3P = CCH + H_2O \longrightarrow (Me_2N)_3PO + MeOH + MeOOCCH \Longrightarrow CHCOOMe$$

$$6 OMe COOMe$$

In presence of methanol, ylids 5-10 can undergo an addition reaction analogous to that described by Schmidbaur (3) to give a phosphorane.

When the trapping reagent used in reaction A is phenol, we obtain a quantitative yield of ylid. Between 0° and 20° this ylid undergoes a complete rearrangement leading to a phosphorane (path c, scheme 2)

OME COOME

OPh

CH

COOME

COOME

$$\delta^{31}$$
P 70.5 ppm

$$\delta^{31}$$
P-52 ppm $J_{HC=CP}$ 23.2Hz

(E configuration)

The carbanionic species $\underline{11}$ (scheme 2) can be trapped with other reagents. For instance, carbon dioxide and phthalimide work well.

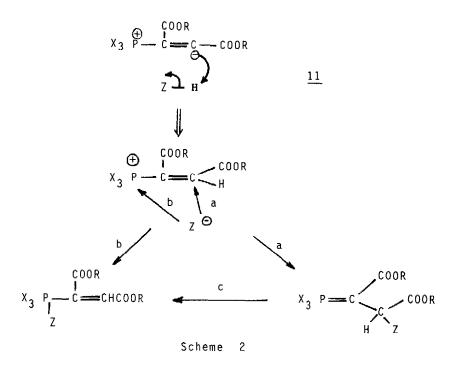
PHOSPHORANES

R	=	-	Ç	=	СН	COOMe				
COOMe										

N°		δ ³¹ p ppm	MeOP		<u>H</u> C=C-P		Me-C	сн ₂ о	COOMe
				J _{HCOP}	δ	J	δ	δ	δ
3	OMe P - R OMe	-36		12.75		45.6	-	3.84 3.73	3.69 3.73
4	OMe P - R OMe	-50	3.45	13.12	6.24	43.5	1.2	1	3.65 3.66

The coupling constant $J_{HC=CP}\simeq 45$ Hz supports the Z configuration This structural relationship is in good agreement with the coupling constants of vinylphosphonates (7).

TABLE II		VIIDS			R = = C < COOMe $C(H) (OMe) COOM$				
TABLE II		YLIDS			- A	R = =C C(H)(OMe)			
		δ ³¹ P.	HCC = P		COOMe	OMe	PNMe ₂		
N°		ppm	δ	J	δ	δ	δ	J	
5 ~	(MeO) ₃ P = R	52	4.8	30	3.5 3.7	3.3	-	_	
6	$(Me_2N)_3P = R$	66	4.5	25.5	3.27 3.48	3.6	2.64	10.5	
7	o NMe ₂ P = R	78.5	4.6	27.7	3.5 3.6	3.3	2.76	10.5	
8	NMe ₂ P = R	64.5	4.65	26	3.45 3.54	3.21	2.73	10.5	
9	Ph NMe 2 N P = R	67	4.65	18	3.55 3.63	3.46	2.85	10.5	
<u>10</u>	$\int_{0}^{0} \int_{P=R}^{OtBu}$	57	4. 50	24	3.48 3.63	3.24	_	-	



During our study, we never observed the Arbuzov reaction which would be the reaction of Z^{Θ} with X_3 (X=0Me). This reaction occurs when the carbanionic intermediate 11 is formed in the absence of trapping reagent (2).

The unstable ylid $\underline{10}$ (Table II) decomposes quickly at room temperature to give a phosphonate.

Reactions of trivalent phosphorous compounds (essentially Ph₃P) with dimethylacetylene dicarboxylate have been the object of intensive study in recent years (4-6). To our knowledge, the ylids and phosphoranes are new (Table I and II). Further studies are in progress to investigate the reactivities of these novel ylids, involving new trapping reagents with synthetic potentialities.

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